

**SYLLABUS – CHEM 223 – Organic Chemistry A – 1<sup>st</sup> Semester – MWF Lecture**  
Fall 2015 - LOYOLA UNIVERSITY CHICAGO (LUC)

**Lecture:** #4666    **Section:** 011    **Mon+Wed+Fri**    **10:25 – 11:15 a.m.**    **Flanner Auditorium 133**

**Discussion:**    #4667    **Section:** 012    Wednesday    11:30 a.m.    Flanner 007  
                  #4714    **Section:** 013                            12:35 a.m.  
                  #4715    **Section:** 014                            1:40 p.m.

**Lecturer:**    Dr. C. Szpunar  
                  Office: Flanner Hall **200B**                            Contact: best in person, 773-508-3128, cszpuna@luc.edu  
                  Emergency Message via Chemistry Dept. Office, 773-508-3100,    fax: 773-508-3086  
                  Office Hours: **Mon** and **Fri:** 11:45 a.m.-1 p.m., **Tues:** 10 a.m.- noon \*\*\* and as arranged **prior**

**Required:**

1. Organic Chemistry, Wade 8<sup>th</sup> ed. (**red**), Prentice Hall, 2013 (ISBN 978-0-321-76841-4) \*or\* Wade, 7<sup>th</sup> ed. (**navy blue**), Prentice Hall, 2010 (ISBN 978-0-321-59231-6) \*or\* 6<sup>th</sup> ed. (**black**), Prentice Hall, 2003 (ISBN 0-13-147871-0) \*or\* 5<sup>th</sup> ed. (white) \*or\* equivalent  
**\*\*\*NO ACCESS CODE NEEDED, NOR DESIRED, NOR SUGGESTED\*\*\***
2. Study Guide and Solutions Manual, Wade & Simek, 8<sup>th</sup> ed. (ISBN 978-0-321-77389-0) \*or\* 7<sup>th</sup> ed. (ISBN 978-0-321-59871-4) \*or\* 6<sup>th</sup> ed. (ISBN 0-13-147882-6) \*or\* 5<sup>th</sup> ed., as appropriate to accompany text

**Suggested / Recommended Materials:**

1. Molecular modeling kit, Darling, Prentice-Hall, Freeman (Maruzen), Proteus, or equivalent
2. Spiral or bound notebook for homework problems

**Optional Materials** (found helpful by some students, but hold off initially – do not purchase immediately):

1. Organic Chemistry as a Second Language, I, Klein (2nd edition, 2008), Wiley (ISBN 978-0-470-12929-6)
2. Barron's Orgo Cards: Organic Chemistry Review, Wang, Razani, Lee, Wu, and Berkowitz (ISBN 0-7641-7503-3) \*or\* Organic Chemistry Study Cards, R Van De Graaff, K Van De Graaff, and Prince, Morton Publishing, 2003 (ISBN 0-89582-577-5) \*or\* equivalent
3. Organic Chemistry: A Short Course, Hart, Craine, Hart, and Hadad (12th edition, 2007), w/CD-ROM, Houghton Mifflin (ISBN-10: 0-618-59073-0, ISBN-13: 978-0-618-59073-5) (*Please see instructor BEFORE purchasing - only for students who may require an alternative-study approach.*)

**Grading** (weighting below) with approximate curved grade guidelines: > 90% A; 75-90% B; 55-75% C

**EXAMS** – 3 – dates scheduled, announced (subject to change, although unlikely) **NO MAKE UPS** **45%**

- UNEXCUSED ABSENCES merit a zero score.
- EXCUSED ABSENCES are handled on a case-by-case basis; grade weighting may be adjusted, depending on the circumstance(s); however, an excused absence **MUST BE CORROBORATED and DOCUMENTED**, e.g., accompanied by a note from the doctor, dentist, hospital rep, or funeral director; by a court summons, plane ticket stub, hospital release form, obituary, or other. **With proper documentation**, religious observance, representing the university, or personal emergency constitutes an Excused Absence.

**QUIZZES** – TBD – **unannounced** (during lecture or discussion or as take-home, **NO MAKE UPS**) **15%**

**HOMEWORK** - assigned per topic / chapter, see Suggested Homework Assignment, below **5%**

**FINAL EXAM** – date scheduled and announced (scheduled by CAS) **35%**

\*\*\* Please note that because this course, *Organic Chemistry*, is cumulative, comprehensive, and improvement-based, and because the final exam is deemed a culminating measure of a student's progress, any student meriting an F on the final exam may achieve a recorded course grade no higher than D, despite total points; a final-exam score of D may merit a course grade no higher than C, despite total points; and a final-exam score of C may merit a course grade no higher than B, despite student's standing otherwise (i.e., despite total points.)

**Course Objective:** To guide, encourage, and foster the learning and understanding of Organic Chemistry – nomenclature, structures, properties, mechanisms, syntheses, and spectroscopy – by the individual student, helping him/her to connect, extrapolate, integrate, and apply the many different aspects learned.

**Student Outcomes:** If successful, the student will learn how to ...

1. identify the various classes of organic compounds, their methods of preparation, and typical reactions.
2. name and draw specific organic compounds.
3. postulate a logical reaction mechanism for simple organic reactions.
4. discriminate amongst relative stabilities of reaction intermediates.
5. plan and write out multi-step syntheses using known reagents / conditions to transform functional groups.
6. prepare for basic purification/separation techniques of organic compounds required in the laboratory.
7. analyze and interpret data from various instruments used in separating and identifying organic compounds: IR, NMR, and UV-vis spectrophotometers and mass spectrograph.

**Lecture and Discussion – Attendance and Attention:** Important and required. Feel free to bring your books and modeling kit to class. Better yet, use them. Prepare for lecture by prior scanning of new material. Come prepared for discussion; be ready to ask questions on assigned homework or yet-unassimilated lecture material.

**Cell Phones: NONE.** Please be courteous and respectful of others. Silent mode during lecture and discussion. **Not allowed in sight or within hearing during exams, subject to confiscation.** NO phone conversations in lecture hall or in discussion class – before class, during class, after class – AT ANY TIME! NO texting – before class, during class, after class – AT ANY TIME! If you must talk or text, take it outside!!!

**Photography: NONE.** No photography of posted quiz or exam keys. No photography of discussion or lecture blackboard or whiteboard content.

**Recording: NONE.** No recording of lectures.

**Academic Honesty:** Essential, expected, and enforced. Dishonesty dictates consequences which may include: (1) notification of Chemistry Department Chair, student's Department Chair, and CAS Dean, (2) documentation in the student's official university record, and (3) dismissal from the university. Immediate consequences will include a **ZERO** on any item in question (quiz or exam). Please refer to the LUC Undergraduate Handbook on policies or the CAS website: [http://www.luc.edu/academics/catalog/undergrad/reg\\_academicintegrity.shtml](http://www.luc.edu/academics/catalog/undergrad/reg_academicintegrity.shtml).

**Study Strategies and Suggestions:** One may approach the study of Organic Chemistry in a manner similar to tackling a new foreign language. Its study will provide a basis to understanding future material – *building constantly, incessantly, and relentlessly* on the structural and mechanistic information presented previously and, hopefully, acquired by the student. Over two semesters, the course will cover: bonding, functional groups, families of aliphatic and aromatic compounds, nomenclature, structures, stereochemistry, reaction mechanisms, multi-step syntheses, and spectroscopic techniques. Because the course is cumulative and builds heavily on prior material, the best plan is to study Organic Chemistry regularly, every day, similar to practicing the piano. Collaboration on homework problems is encouraged, especially in a timely fashion. Experience dictates that positive outcomes (for exam and course grades) are directly proportional to working and understanding the assigned problems on a regular basis, i.e., applying the concepts learned to non-generic situations.

Typically, Organic Chemistry is not efficiently self-taught. Overnight cramming will probably not produce success! The student should quickly read the chapter/segment to be covered BEFORE lecture to improve lecture comprehension. After lecture, careful detailed re-reading of the chapter/segment and focused working of the assigned problems are appropriate, necessary, essential, and expected. In addition to student's participation in lecture, discussion, reading, and homework, joining and contributing to a study group is strongly encouraged.

*If anticipating a passing grade of C, the minimal time per week devoted to Organic Chemistry is estimated at 4 hr for lecture and discussion, 4-10 hr for reading, and 4-10 hr for homework.*

**Homework Package:**

\*\*\* due Wednesday, Nov. 11, 2015 by 2 p.m. \*\*\*

Hand in at least 15 completed problems (or parts of multiple problems) per chapter, **Chapters 1–10**. (Underlined problems may be particularly insightful or anticipatory.) If not using a notebook, include your name and the date worked on **each** page. **Clearly mark the chapter** and problem number for full credit. Be sure to **staple/clip** packets in chapter order \*\*\*\*and/or\*\*\*\* **flag each individual chapter in packet/notebook** (to be returned to the student). **Deliver at lecture or to Flanner 200B**, directly to the lecturer, NOT to the dept. office!!!

**Suggested Homework Assignment** (for Wade's 8<sup>th</sup> edition):

- Chap 1: 2-4, 6-10, 14-15, 17-18, 21, 23, 25, 27, 29, 31-32, 34-36, 39-40, 42-45, 52, 55
- Chap 2: 3-5, 7-11, 15-22, 35-36, 39-42, 44 (note solutions manual p.40 – functionality)
- Chap 3: 1-8, 10, 14-17, 20-21, 24-25, 29-30, 32-35, 37, 39, 44, 46
- Chap 4: 1-2, 4, 7, 9-13, 18-21, 28-32, 35-39, 40, 42, 45-46
- Chap 5: 1-7, 14-22, 25-31
- Chap 6: 1-3, 5-7, 9-12, 14-20, 22-26, 27 (b-e), 29-45, 53, 56
- Chap 7: 1-2, 4-8, 11-13, 15-16, 19, 21-25, 30-36, 38, 40-46
- Chap 8: 1-2, 4-19, 21-22, 26-29, 32-37, 45-47, 49, 50, 63  
(N.B. solutions manual p.178 – addition + stereochemistry)
- Chap 9: 1-2, 5-13, 18-20, 23-30, 32-37
- Chap 10: 1-4, 7-10, 13-21, 23-26, 30-33, 37-41, 43, 49, 51
- Chap 11: 1-2, 4-6, 9-13, 21-22, 26, 28-31, 37, 39-44, 46
- Chap 12: 2-7, 11-12, 14-16, (17-fragmnt), 24-25
- Chap 13: 2-11, 13-15, 22-25, 27, 32-36, 38-40, 42-44, 47-48
- Chap 14: 4, 6, 8-10, 12, 14-15, (18 - mustard gas, bleach), 20, 22, 25, 27-33, (34), 39

(for Wade's 7<sup>th</sup> edition):

- Chap 1: 2-4, 6-10, 14-15, 17-18, 20-21, 23, 25, 27, 29, 31-32, 34-36, 39-40, 42-45, 54
- Chap 2: 3-5, 7-11, 15-23, 35-36, 39-42, 44 (note solutions manual p.38 – functionality)
- Chap 3: 1-5, 9-10, 14-17, 20-21, 24-25, 29-30, 32-35, 37, 39, 44, 46
- Chap 4: 1-2, 4, 7-13, 18-22, 26, 28-31, 34-39, 42, 45-46, 50
- Chap 5: 1-7, 15-22, 25-31
- Chap 6: 1-3, 5-7, 9-12, 14-16, 19-20, 22-24, 30-38, 40-45, 53, 56
- Chap 7: 1-2, 4-8, 11, 13, 15, 19, 21, 23-25, 30-36, 38, 40-46
- Chap 8: 1-2, 4-15, 17-19, 21-22, 27-29, 32, 34-37, 46-47, 49, 63-64  
(N.B. solutions manual p.168 – addition + stereochemistry)
- Chap 9: 1-2, 5-8, 12, 18-19, 23-29, 33-34, 36-37
- Chap 10: 1-4, 7-10, 13-20, 23-26, 30-33, 37-39, 49, 51
- Chap 11: 1-2, 5-6, 9-13, 21-22, 26, 31, 34, 39-45
- Chap 12: 2-7, 11-12, 14-16, (17-fragmnt), 24-25
- Chap 13: 2-11, 14-15, 22-25, 32-36, 38-44
- Chap 14: TBD by student

(for Wade 6<sup>th</sup> edition):

- Chap 1: 2-4, 7-11, 13-15, 17-18, 20, 23, 25, 27, 29, 32, 34-36, 39-40, 42, 44-45, 47
- Chap 2: 5, 7-11, 15-23, 35-36, 39-42, 44
- Chap 3: 1-5, 9-10, 14-15, 20-21, 29-35, 37, 39, 44, 46
- Chap 4: 1-2, 4-13, 18-22, 26, 28-31, 34-39, 42, 45-46, 50
- Chap 5: 1-3, 5-8, 16-22, 25-30
- Chap 6: 1-7, 11-12, 14-16, 19-20, 22-24, 31-37, 44-45, 53, 56
- Chap 7: 1-2, 4-5, 7-8, 13, 15, 19, 21, 23, 25, 34-35, 38, 40, 42-45
- Chap 8: 1-2, 4-6, 8-15, 17-19, 27-28, 32, 34-37, 47, 49, 63-64
- Chap 9: 1, 2, 5-8, 12, 18-19, 23-24, 33-34
- Chap 10: 3, 7, 8-10, 13-15, 17-20, 24-26, 37-39, 49
- Chap 11: 2, 5-6, 9-11, 13, 22, 31, 34, 41-45
- Chap 12: 2-12, 14-19, 23, 25
- Chap 13: 2-36, 38-45
- Chap 14: TBD

**Lecture Outline (subject to change) – Fall 2015**

<u>Week</u>	<u>Date</u>	<u>Chap-Lecture</u>	<u>Topic</u>	<b>***EVENT***</b>
1	Aug 24 26 28	1-1 1-2 2-1	Intro: Lewis structures, bonding, resonance, acid-base, nomenclature ... Structure and properties	
2	Aug 31 Sept 2 4	2-2 3-1 3-2	Alkanes, cycloalkanes, bicyclics	
3	Sept 7 9 11	*** 4-1 4-2	***** <b>Monday</b> ***** Chemical rxns - free radical halogenation, kinetics, intermediates ...	<b>LABOR DAY – HOLIDAY ***</b>
4	Sept 14 16 18	4-3 5-1 5-2	Stereochemistry – chirality, isomers ...	
5	<b>Sept 21</b> 23 25	<b>5-3</b> 6-1	Alkyl halides, nucleophilic substitution and elimination	<b>***** MONDAY ***** EXAM I (Chapters 1-4)</b>
6	Sept 28 30	6-2 6-3		
7	Oct 2 5/6 7 9	6-4 *** 7-1 7-2	Alkenes	<b>***** Mon/Tues ***** MIDTERM BREAK *****</b>
8	Oct 12 14 16	7-3 8-1 8-2	Alkenes – rxns	
9	<b>Oct 19</b> 21 23	<b>8-3</b> 9-1	Alkynes	<b>***** MONDAY ***** EXAM II (Chapters 5-7)</b>
10	Oct 26 28 29 30	9-2 9-3 *** 10-1	Alcohols	<b>***** Thursday ***** (last day to withdraw with a W) *****</b>
11	Nov 2 4 6	10-2 10-3 11-1	Alcohols – rxns	
12	Nov 9 11 13	11-2 11-3 12-1	Spectroscopy – IR and MS	<b>*** Wednesday *** HW assignment due for grade *****</b>
13	<b>Nov 16</b> Nov 18 20	<b>12-2</b> 12-3		<b>***** MONDAY ***** EXAM III (Chapters 8-11)</b>
14	Nov 23 25-27	13-1 ***	Spectroscopy – NMR	<b>***** Wed-Fri ***** THANKSGIVING DAY – HOLIDAY *****</b>
15	Nov 30 Dec 2 4	13-2 13-3 14-1	Ethers, epoxides, sulfides (student to finish on his/her own, if time does not permit lecture coverage)	
16	<b>Dec 7, Mon</b>		<b>Cumulative FINAL EXAM, 9:00 – 11:00 a.m., Flanner Auditorium</b>	